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SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how

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many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

7 Chemical Formulas and Chemical Compounds

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STUDY. PLAY. The basis of an ionic bond is the ___ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a ___ noble gas. Ions that are formed from more than one atom are called ___

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study the thing. Terms in this set (25)
Chemical formula. indicates the relative number of atoms of each kind in a chemical compound. Reveals the number of atoms of each element contained in a single molecule ...

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Modern Chemistry 35 Chapter Test
Name Class Date Chapter Test B,
continued Write the orbital notation
for the following elements in the
space provided Modern chemistry
chapter 7 test answer key. 37. lithium,

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atomic number 3 38. carbon, atomic number 6 39 Modern chemistry chapter 7 test answer key. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and ...

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Answers

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(c) the charge on each Fe ion. 7
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Modern Chemistry CHAPTER 7
HOMEWORK 7-5 VOCABULARY

Complete each sentence. 1. Oxidation numbers are assigned to the atoms composing a compound or ion in

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STUDY. PLAY. Formula Masses. any
molecule, formula unit, or ion is the

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sum of the average atomic masses of all atoms represented in its formula. a compound's molar mass is numerically ... Modern Chemistry Chapter 7 Section 1 Review. 4 terms.

Modern Chemistry Chapter 7 Section 3

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Review Answers

elements in the space provided

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answer key. 37. lithium, atomic

number 3 38. carbon, atomic number

6 39 Modern chemistry chapter 7 test

answer key. neon, atomic number 10

PART VI Write the answers to the

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questions on the line to the left, and
... Modern Chemistry Chapter 7 Test
Answer Key

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Answers

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